

# Volcano in the lab: a wax volcano in action: teacher's notes

## Level

This activity is designed for students aged 11-14, as a simple demonstration of igneous activity.

It can also be used with students aged 14-16 when discussing of the structure of the Earth and the physical properties of its layers.

## Topic

The aim of this topic is to simulate ways in which both **extrusive** and **intrusive igneous** rocks may form.

## Description

This activity consists of a teacher-led demonstration for the whole class in which layers of sand and wax in a beaker of water are used to model how **igneous** rocks form both underground and at the surface. It may be that some teachers would wish to allow students to carry out the practical under very close supervision.

## Context

**Volcanoes** are exciting – hence all the volcano footage on TV. They can be used to fire students' imaginations, and safe analogues of the behaviour of molten rocks can be demonstrated in the school laboratory.

Students will have seen TV coverage of volcanic eruptions, and may even have spent holidays in volcanic regions. They will also know that temperatures generally increase with depth in the Earth.

At the end of the activity students should appreciate how rising **magma** can cool at and below the Earth's surface, forming rocks which we call **igneous**.

## Teaching points

The demonstration can follow the showing of selected video clips of volcanic eruptions such as those available at <http://volcano.und.nodak.edu/vwdocs/movies/movie.html>.

It is a common misconception that there is a universal layer of molten rock lying just below the Earth's crust. This imaginary layer is often erroneously equated with the **mantle**, which is, in fact solid. Localised heating, and / or reduction in pressure, lead to partial melting, but the **magma chambers** which form are only tens of kilometres across, not mantle-wide. Students also find it difficult to visualise that some molten rock can set below the Earth's surface to form **intrusive igneous rocks**.

The reason why temperature increases with depth in the Earth is mainly because of radioactive decay of minerals within the Earth, and the fact that the hundreds of kilometres of overlying rock provide a very good insulator. During radioactive decay of an element, new elements and sub-atomic particles are formed. The total mass of these is very slightly less than the mass of the original element and the difference ( $\Delta m$ ) is converted into the equivalent amount heat energy ( $E$ ) in line with the equation  $E = \Delta mc^2$ , where  $c$  is the speed of light.

## Timing

The demonstration itself takes about 10 minutes, with discussion to follow.

## Apparatus

- one 500 cm<sup>3</sup> or 600 cm<sup>3</sup> Pyrex™ beaker
- Bunsen burner
- heat proof mat
- tripod
- gauze
- safety screen

## Chemicals

- red candle wax
- washed sand (sand can be washed by putting some in a bucket and using rubber tubing to run water run into the bucket and allowing the water to overflow into a sink until it runs clear)

## Safety notes

- Wear eye protection.
- The activity is safer than it sounds - the only potential hazard is a cracked beaker, when some localised spillage of hot wax can occur: the water remains cold throughout.
- It is the responsibility of the teacher to carry out an appropriate risk assessment.

## The demonstration

Melt red candle wax into the base of the beaker to about 1 cm depth. Cover this with a layer of sand about 1 cm thick above the wax. Add water to fill the beaker about three quarters full.

Apply a strong source of heat to one part of the base of the beaker and let it cool to form a solid layer. Students need to concentrate because the 'eruption' often happens without much warning, other than an ominous crackling sound as the wax melts! The heat source is removed whilst there is still some wax left on the bottom of the beaker since this allows 'lava tubes' and intrusions to form more effectively.

Figure 1 shows an example of what might be observed.

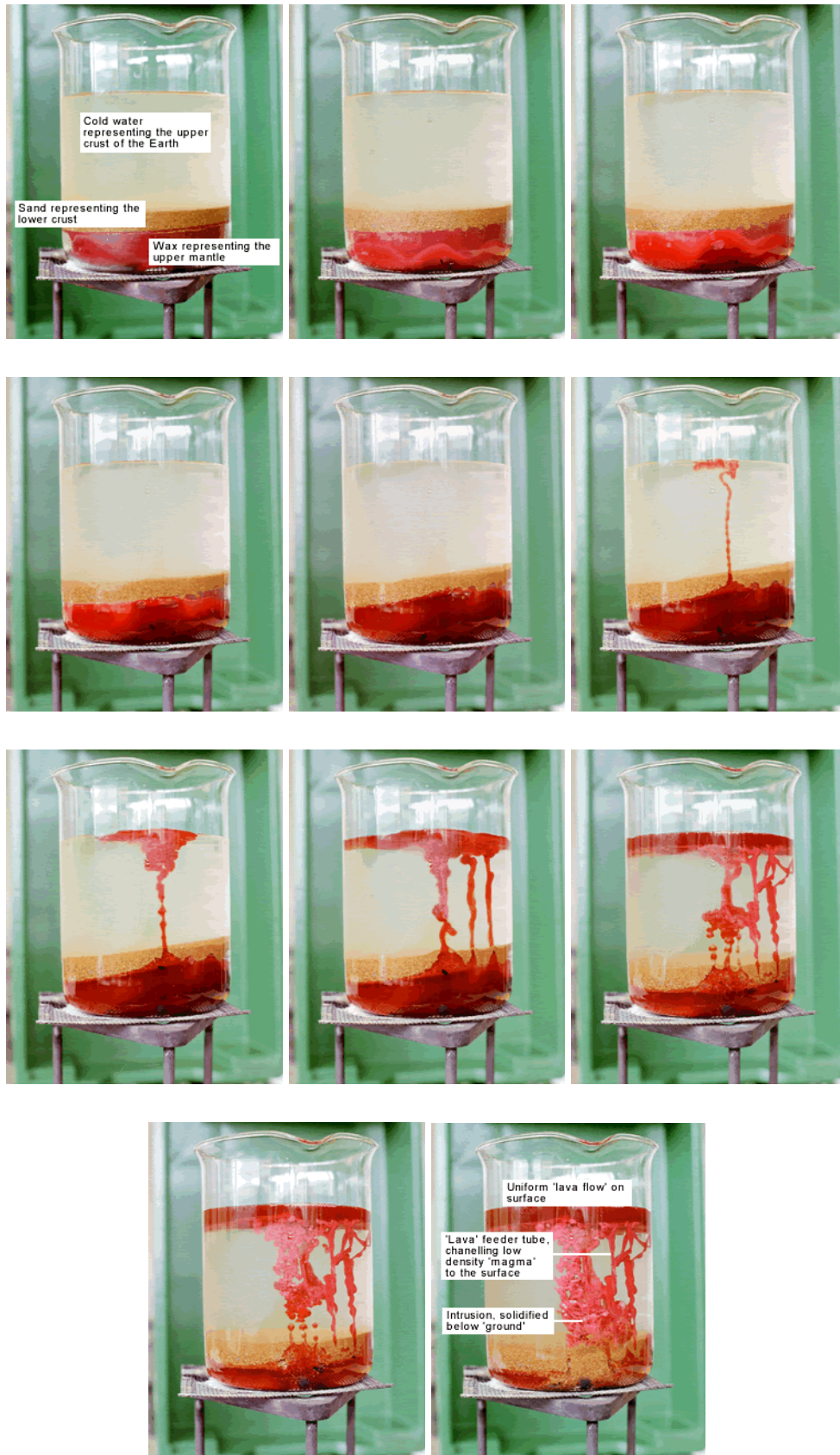


Figure 1 A volcano in the lab

## Points to bring out

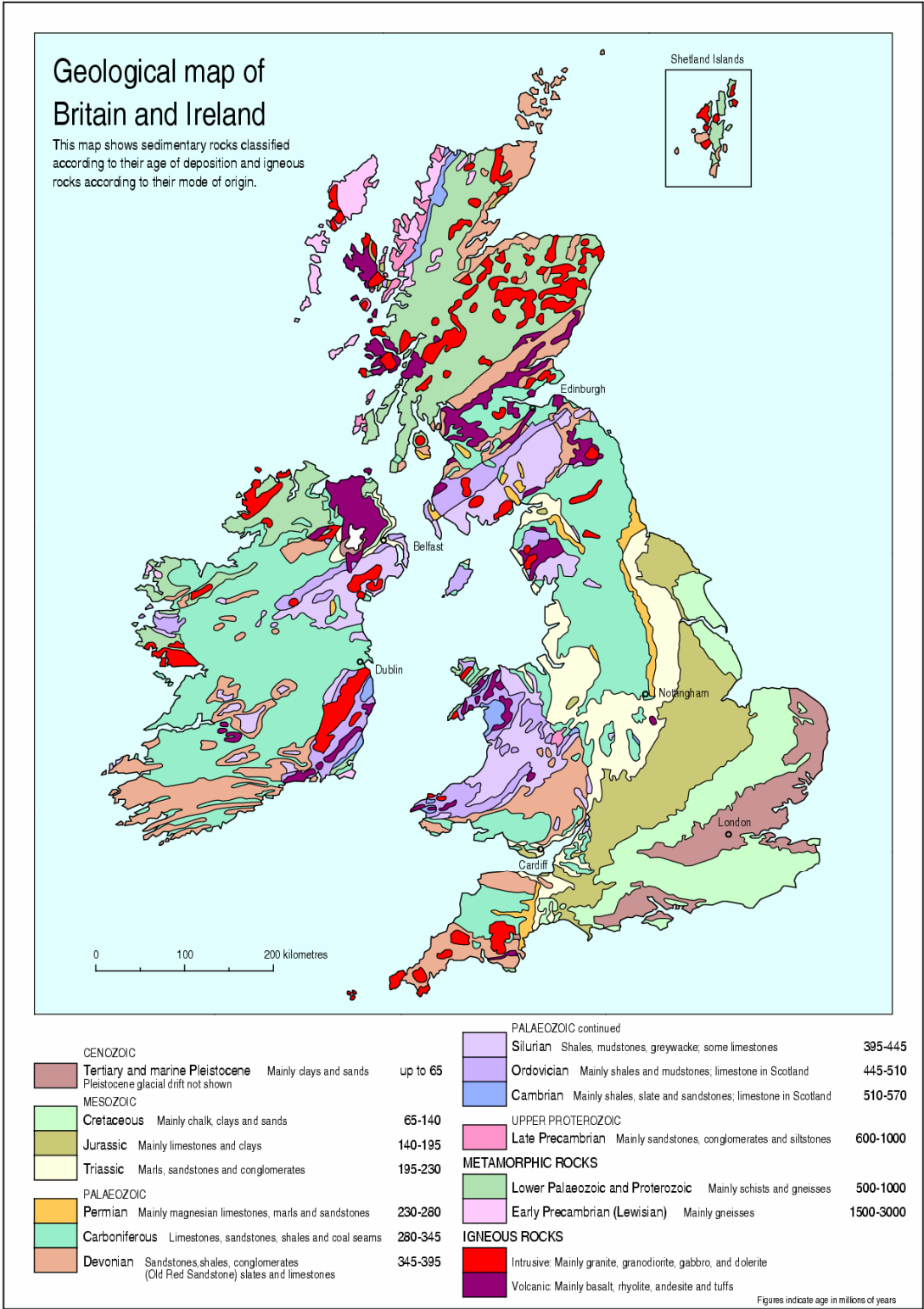
- Both the sand *and* the water represent the **crust** of the Earth – the water does *not* represent the sea.
- The wax layer represents a layer in the Earth below the **crust** (called the **mantle**).
- The mantle is solid. At certain points it becomes hot enough to melt.
- When the wax melts, it rises because of its lower density. It represents molten rock, known as **magma**.
- Some of the wax rises rapidly to the surface, imitating a volcanic eruption. It is very runny and spreads out evenly over the surface of the water (usually). This represents the way in which some **lavas** may cover huge areas, arising from fissure eruptions, which produce a greater total amount of lava than the better-known individual volcanoes.
- Some of the wax can be seen rising through 'tubes' of wax which insulate it from the surrounding cold water and enable it to reach the surface. The location of this tube could be linked to a point above a 'weak spot' in the rock layer, *ie* the tube forms above this weakness finding the easiest way to the surface. This happens in Nature too. Students may be surprised that such a vast amount of magma has passed so quickly through such a small tube.
- Some of the wax sets very quickly in the cold water, forming grotesque shapes. These represent **intrusive igneous rocks**. Once the wax has all set, the 'lava layer' may be removed and the water poured off in order to study the shapes of the '**intrusions**'. This is analogous to the removal of layers of rocks by weathering and erosion.
- Reference can be made to the geological map of Great Britain and Ireland, Figure 2. Widespread sheets of lava form the Antrim plateau in Northern Ireland: masses of intrusive igneous rocks are shown as big red blobs in Devon and Cornwall, Southern Uplands of Scotland *etc*.
- Students can be challenged to say which aspects of the model are not consistent with the natural world. The most important one is that the surface eruption sets very slowly, whilst the '**intrusions**' set very quickly. In reality, the reverse would be true, because of the higher ambient temperatures at depth and the insulating properties of several kilometres of rock. **Lavas** may become solid within days, months or years, whereas a deep-seated intrusion of several tens of cubic kilometres may take millions of years to cool to the ambient temperature. Of course, the wax merely sets: it does not form crystals.
- Students can be reminded that slow cooling leads to large crystals.

## Extension

In reality, complete melting of rocks below ground is seldom achieved. Rocks partially melt, and the minerals of lowest melting points are the ones which melt and rise (they are also the least dense minerals). This can be shown by preparing a mixture of chopped wax and gravel in a metal container. When heated in front of students the wax melts and rises, whilst the gravel does not. It is possible to do this at the same time as the volcano demonstration, but experience shows that glass beakers tend to be more susceptible to cracking in this experiment.

## **Acknowledgements**

The original idea for the wax volcano model came from Mike Tuke and is described in M. Tuke *Earth Science Activities and Demonstrations*, London: John Murray, 1991.



**Figure 2 Geological map of Great Britain and Ireland**